

BBF-014-1041007

Seat No.

B. Pharm (Sem. I) Examination

July - 2021

Pharmaceutical Inorganic Chemistry (BP 104T)

Faculty Code: 014

Subject Code: 1041007

Time: 3 Hours [Total Marks: 75

1 Answer the following questions.

 $10 \times 2 = 20$

- (a) Give difference between poison and antidote.
- (b) Write the composition of ORS.
- (c) Differentiate the term : hypochlorhydria and hyperchlorhydria.
- (d) What are gastrointestinal agents? Classify them with suitable Examples.
- (e) Define the term: Buffer capacity?
- (f) What do you mean by Astringent with its two example?
- (g) What do you mean by emetics? Explain it with suitable examples.
- (h) Explain properties of α, β, γ radiations.
- (i) What are expectorants? Give suitable examples of this category.
- (i) What are Dentifrices.
- 2 Answer any two out of the following:

 $2 \times 10 = 20$

- (a) Explain physiological acid base balance.
- (b) Write a note on limit test of Arsenic.
- (c) Define: Impurity. Classify it with suitable examples. Enumerate sources of impurities and explain any one in detail.

- 3 Answer any Seven out of the following: 7×5=35
 - (a) Discuss the role of fluoride in the treatment of dental caries. Give preparation and use of sodium fluoride.
 - (b) What are antimicrobials? Classify them. Explain its mechanism of action.
 - (c) Define: Haemantinics. Give suitable examples of this category. Explain assay principle of ferrous sulphate.
 - (d) What are the ideal properties of antacids? Give assay principle and method of preparation of sodium bicarbonate.
 - (e) Explain cyanide poising and its treatment.
 - (f) What is half life? Write a note on GM counter.
 - (g) Give preparation, properties, assay and uses of sodium thiosulphate.
 - (h) Explain limit test of Chloride and Sulphate.
 - (i) Give assay principle of:
 - (a) Ammonium chloride (b) Hydrogen peroxide.

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